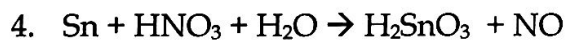
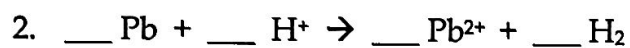
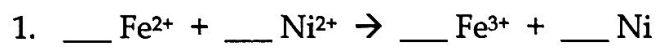
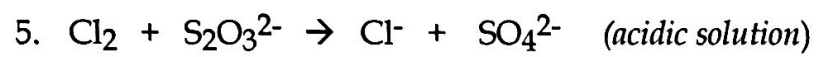


SUPPLEMENTAL PRACTICE: BALANCING REDOX REACTIONS WORKSHEET

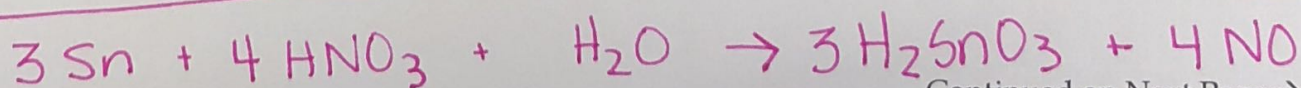
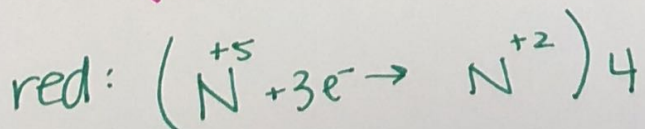
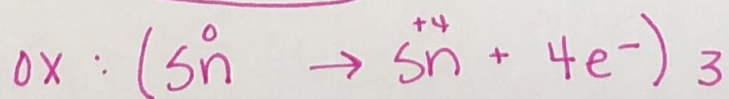
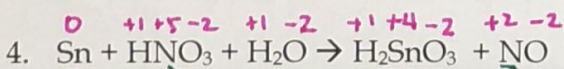
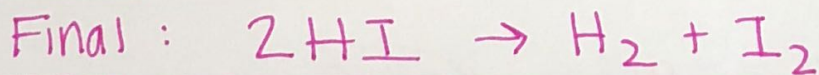
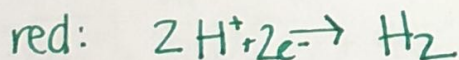
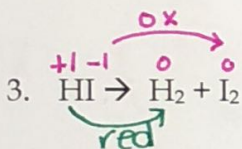
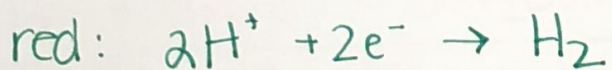
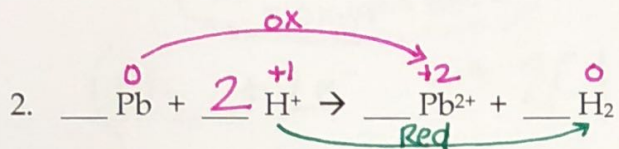
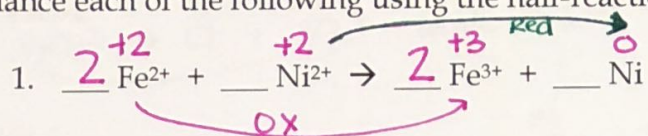
Balance each of the following using the half-reaction method.



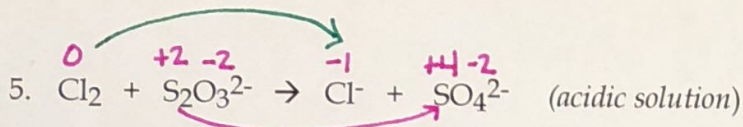


SUPPLEMENTAL PRACTICE: BALANCING REDOX REACTIONS WORKSHEET

Balance each of the following using the half-reaction method.



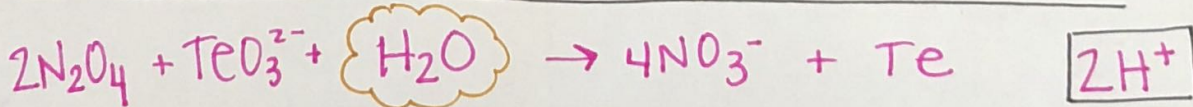
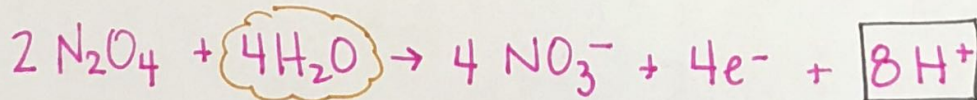
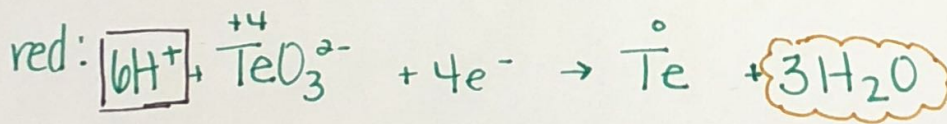
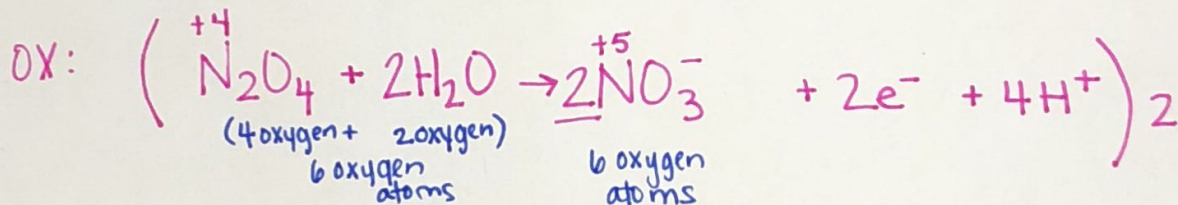
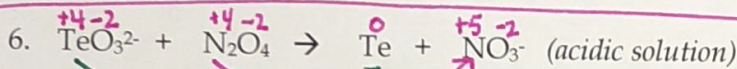
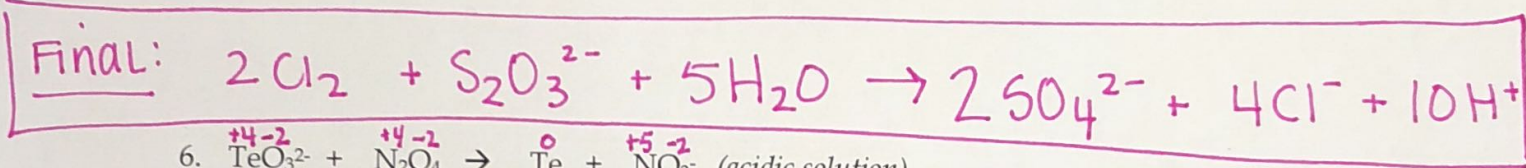
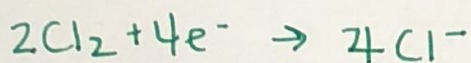
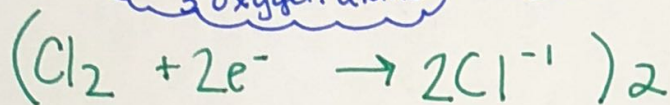
Continued on Next Page →



only 5 H₂O added since S₂O₃²⁻ has 3 oxygen atoms

There are 8 oxygen atoms

red:



rewrite

CANCEL LIKE TERMS IN THE HALF-REACTIONS