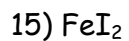
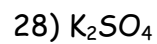
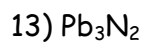
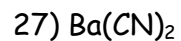
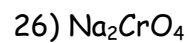
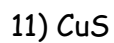
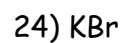
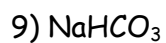
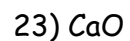
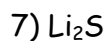
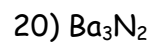
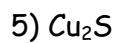
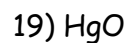
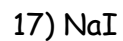
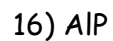


IONIC COMPOUNDS - FORMULAS AND NOMENCLATURE WORKSHEET

NAME: _____

Write the correct name for:



Write the correct formula for:

1) magnesium oxide

2) lithium bromide

3) calcium nitride

4) aluminum sulfide

5) potassium iodide

6) strontium chloride

7) sodium sulfide

8) beryllium bromide

9) sodium oxide

10) calcium fluoride

11) boron phosphide

12) aluminum oxide

13) iron(II) chloride

14) copper(I) sulfide

15) lead(IV) iodide

16) tin(II) fluoride

17) mercury(I) bromide

18) tin(II) oxide

19) chromium(III) oxide

20) gold(I) iodide

21) manganese(II) nitride

22) cobalt(III) phosphide

23) silver carbonate

24) potassium hydrogen phosphate

25) aluminum hydroxide

26) sodium hydrogen carbonate

27) calcium acetate

28) potassium permanganate

29) calcium perchlorate

30) lithium carbonate

31) magnesium hydrogen sulfite

32) sodium hypochlorite

33) tin(IV) chlorite

34) mercury(II) phosphate

35) tin(II) carbonate

